

8,10-Dibenzylimidazonornbornylene

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Key indicators

Single-crystal X-ray study

$T = 150\text{ K}$

Mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$

R factor = 0.059

wR factor = 0.147

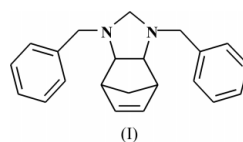
Data-to-parameter ratio = 27.9

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound, 8,10-dibenzyl-1,4-methanobicyclo[4.3.0]-[8,10]diazanon-5-ene, $\text{C}_{22}\text{H}_{24}\text{N}_2$, was obtained through a [4 + 2]-cycloaddition reaction. It crystallizes in the monoclinic space group $P2_1/c$ ($Z = 4$). The 1,3-imidazole ring is *endo* with respect to the norbornylene skeleton. The benzyl protecting groups are situated along the sides of the molecule, towards the back.

Comment

Radioimmunotherapy is a new method for treating certain types of cancer, such as leukemia or lymphoma (Chatal *et al.*, 1993). This approach, which is complementary to conventional treatments, led us to develop new radionuclide chelating agents (Gouin, Gestin, Joly *et al.*, 2002; Gouin, Gestin, Reliquet *et al.*, 2002; Gouin, Gestin, Remaud *et al.*, 2002; Ouadi *et al.*, 2000). Moreover, numerous complexes of lanthanides and DTPA (diethylenetriaminepentaacetic acid) analogues have proved stable enough to be used in a physiological medium as radiopharmaceuticals (Liu & Edwards, 2001; Wu *et al.*, 1997). The most promising results relate to studies aimed at increasing the rigidity of the chelating structure. The introduction of a semi-rigid preformed skeleton minimizes the freedom of donor atoms and thereby has a significant effect on the stability of their metal complexes (Fossheim *et al.*, 1991; McMurry *et al.*, 1998). These considerations led us to synthesize ligands with a rigid norbornane skeleton.



The synthetic strategy was based on obtaining the intermediate title compound, (I), where the two N atoms, useful for further metal chelation, are protected by the benzyl groups. The structure determination was a key element in showing that the two amines are indeed in *endo* positions of the ring. Atom C7 is clearly on the opposite side from atoms N8 and N10 relative to the C1/C2/C3/C4 plane, with C7 0.921 (2) Å above that plane and N8 and N10 -1.199 (2) and -1.185 (2) Å, respectively, below it. The compound shows a staircase structure, with angles of 98.55 (9), 117.81 (9) and 104.20 (8)° for C7-C4-C3, C4-C3-N10 and C3-N10-C9, respectively. The orientation of the benzyl protecting groups minimizes the interactions. The compound was obtained after selective reduction of a carbonyl group in the C9 position by the action of lithium aluminium hydride. This selectivity is confirmed, insofar as the norbornylene double bond is conserved. The C5-C6 bond length is clearly in

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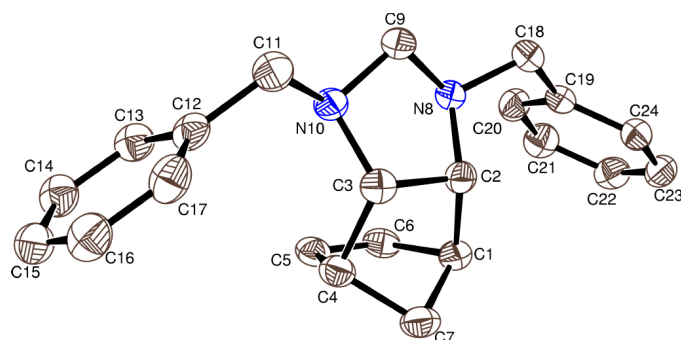


Figure 1
The molecular structure of (I), showing 50% probability displacement ellipsoids. H atoms have been omitted for clarity.

accordance with classical values for a Csp^2-Csp^2 bond [1.3318 (17) Å].

Experimental

The title compound was prepared by mixing 7,9-dibenzyl-2,5-methanobicyclo[4.3.0]-7,9-diazanon-3-en-8-one (0.88 g, 0.00267 mol) with $LiAlH_4$ (1 g, 0.0267 mol) and anhydrous tetrahydrofuran (35 ml) under argon. The mixture was refluxed for 21 h and cooled to 273 K before a 15% solution of sodium hydroxide (2 ml) was slowly added, with continuous stirring, over a period of 10 min at room temperature. The mixture was filtered through a Celite pad and washed with water (200 ml) before the filtrate was extracted with dichloromethane (300 ml). The organic phase was washed with brine, dried ($MgSO_4$), filtered and evaporated. The residue was purified by flash chromatography on silica, using dichloromethane/ethyl acetate (95/5) as eluant. Single crystals suitable for X-ray analysis were obtained by slow evaporation at room temperature from diethyl ether.

Crystal data

$C_{22}H_{24}N_2$	$D_x = 1.190 \text{ Mg m}^{-3}$
$M_r = 316.4$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 17065 reflections
$a = 15.4338 (4) \text{ \AA}$	$\theta = 2.9\text{--}32.0^\circ$
$b = 8.2487 (2) \text{ \AA}$	$\mu = 0.07 \text{ mm}^{-1}$
$c = 13.8748 (3) \text{ \AA}$	$T = 150 \text{ K}$
$\beta = 92.7699 (8)^\circ$	Block, colourless
$V = 1764.32 (7) \text{ \AA}^3$	$0.31 \times 0.23 \times 0.18 \text{ mm}$
$Z = 4$	

Data collection

Nonius KappaCCD diffractometer	$R_{int} = 0.056$
φ and ω scans	$\theta_{max} = 32.0^\circ$
Absorption correction: none	$h = -23 \rightarrow 22$
17647 measured reflections	$k = -12 \rightarrow 11$
6050 independent reflections	$l = -20 \rightarrow 13$
4012 reflections with $I > 2\sigma(I)$	

Refinement

Refinement on F^2	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.059$	$w = 1/[\sigma^2(I) + 0.0025I^2]$
$wR(F^2) = 0.147$	$(\Delta/\sigma)_{max} < 0.001$
$S = 1.45$	$\Delta\rho_{max} = 0.30 \text{ e \AA}^{-3}$
6050 reflections	$\Delta\rho_{min} = -0.25 \text{ e \AA}^{-3}$
217 parameters	

All H atoms were initially found in a difference Fourier synthesis but were fixed in idealized positions in the final refinement. Riding isotropic displacement parameters, with $U_{iso}(H) = 1.2U_{eq}(C)$, were used for all H atoms.

Data collection: *COLLECT* (Nonius, 1998); cell refinement: *HKL SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *HKL DENZO* and *SCALEPACK* (Otwinowski & Minor, 1997); program(s) used to solve structure: *SHELXTL* (Sheldrick, 1995); program(s) used to refine structure: *JANA2000* (Petricek & Dusek, 2000); molecular graphics: *DIAMOND* (Brandenburg & Berndt, 1999); software used to prepare material for publication: *JANA2000*.

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References

- Brandenburg, K & Berndt, M. (1999). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.
- Chatal, J.-F., Peltier, P., Bardies, M., Chetanneau, A., Resche, I., Mahe, M. & Charbonnel, B. (1993). *Med. Nucl. Imag. Met.* **17**, 81–94.
- Fossheim, R., Dugstad, H. & Dahl, S. G. (1991). *J. Med. Chem.* **34**, 819–826.
- Gouin, S. G., Gestin, J.-F., Joly, K., Loussouarn, A., Reliquet, A., Meslin, J.-C. & Deniaud, D. (2002). *Tetrahedron*, **58**, 1131–1136.
- Gouin, S. G., Gestin, J.-F., Reliquet, A., Meslin, J.-C. & Deniaud, D. (2002). *Tetrahedron Lett.* **43**, 3003–3005.
- Gouin, S. G., Gestin, J.-F., Remaud, P., Faivre-Chauvet, A., Meslin, J.-C. & Deniaud, D. (2002). *Synlett*, **12**, 2080–2082.
- Liu, S. & Edwards, S. (2001). *Bioconjugate Chem.* **12**, 7–34.
- McMurry, T. J., Pippin, C. G., Wu, C., Deal, K. A., Brechbiel, M. W., Mirzadeh, S. & Gansow, O. A. (1998). *J. Med. Chem.* **41**, 3546–3549.
- Nonius (1998). *COLLECT*. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr and R. M. Sweet, pp. 307–326. New York: Academic Press.
- Ouadi, A., Loussouarn, A., Remaud, P., Morandea, L., Apostolidis, C., Musikas, C., Faivre-Chauvet, A. & Gestin, J.-F. (2000). *Tetrahedron Lett.* **41**, 7207–7209.
- Petricek, V. & Dusek, M. (2000). *JANA2000*. Institute of Physics, Prague, Czech Republic.
- Sheldrick, G. M. (1995). *SHELXTL*. Version 5.0. Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Wu, C., Kobayashi, H., Sun, B., Yoo, T. M., Paik, C. H., Gansow, O. A., Carrasquillo, J. A., Pastan, I. & Brechbiel, M. W. (1997). *Bioorg. Med. Chem.* **5**, 1925–1934.